

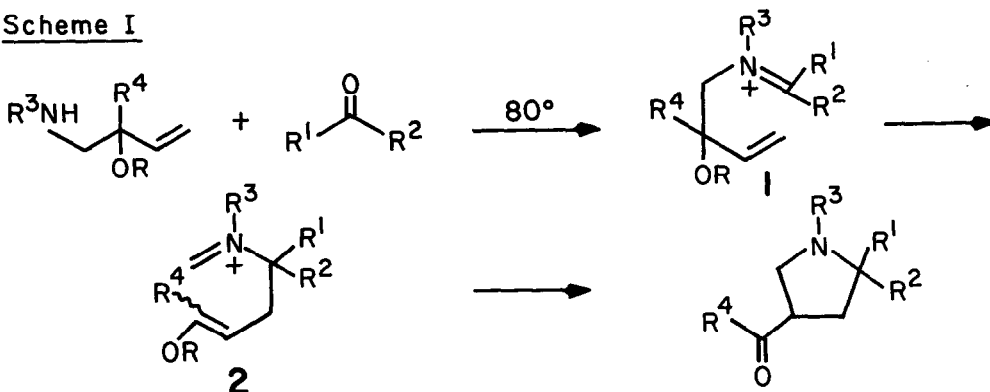
DIRECTED 2-AZONIA-[3,3]-SIGMATROPIC REARRANGEMENTS. A CONVENIENT
PREPARATION OF SUBSTITUTED 1-AZASPIRO[4,5]DECANES

Larry E. Overman,^{*1} Masa-aki Kakimoto, and Makoto Okawara
Department of Chemistry, University of California, Irvine,
California, 92717; and Research Laboratory of Resources
Utilization, Tokyo Institute of Technology, 4259 Nagatsuta,
Midori-Ku, Yokohama, Kanagawa, Japan

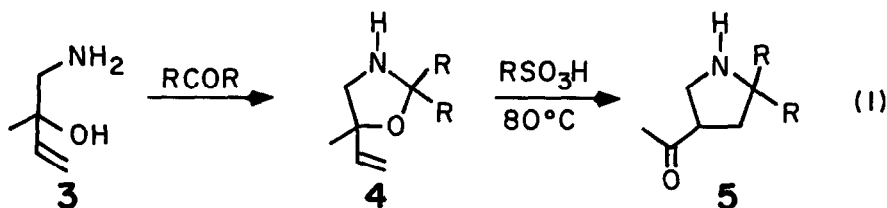
SUMMARY: 3-Acetylpyrrolidines are prepared from 5-methyl-5-vinyloxazolidines in a reaction which involves a directed 2-azonia-[3,3]-sigmatropic rearrangement.

We recently reported² that salts of 2-alkoxy-3-butenamines react with a variety of aldehydes to produce 3-acylpyrrolidines in a single step, and in excellent yield. This pyrrolidine synthesis exploited the ability of a properly placed oxygen substituent to direct the course of the low temperature 2-azonia-[3,3]-sigmatropic rearrangement³ ($\underline{1} + \underline{2}$) by capturing the rearranged sigmatropic isomer in an intramolecular Mannich fashion, (Scheme I, $R^1=H$). This reaction was not

Scheme I

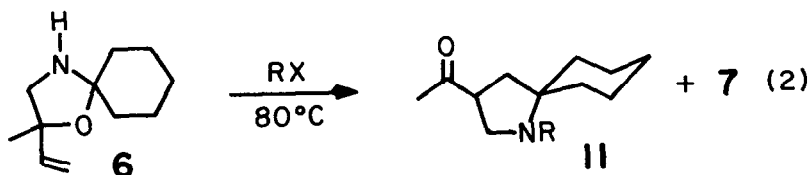


successfully extended to ketones, however, due to the inability to form the starting iminium ion $\underline{1}$ in these cases.² We report in this letter that iminium ion $\underline{1}$ may also be generated from an oxazolidine precursor. This approach (eq 1) allows cyclohexanones to be utilized as the carbonyl components in this pyrrolidine synthesis, and results in a convenient construction of substituted 1-azaspiro [4,5] decanes.



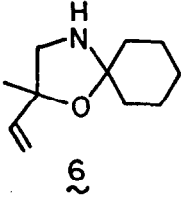
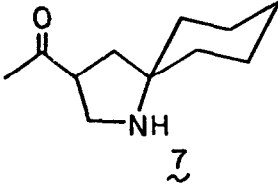
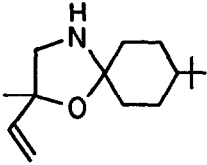
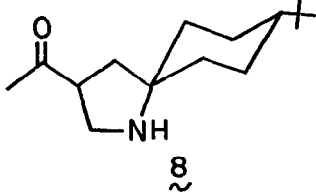
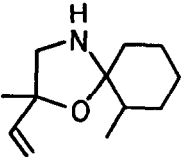
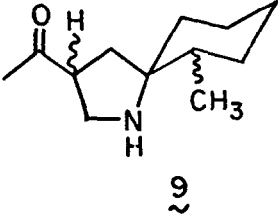
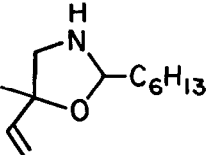
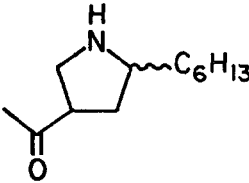
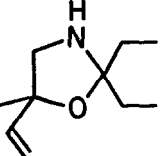
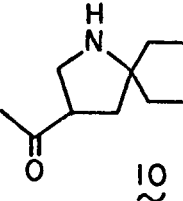
The required 5-vinyloxazolidines were prepared from readily available ⁴1-amino-2-methyl-3-buten-2-ol (3) by azetotropic dehydration in refluxing benzene, or alternatively at room temperature in THF in the presence of Na₂SO₄. The results obtained when oxazolidines 4 were treated with 1 equiv of anhydrous d-10-camphorsulfonic acid in refluxing benzene for 24 h are summarized in Table I. Pyrrolidines 5 were formed in good yields from 5-vinyloxazolidines derived from cyclohexanones and aliphatic aldehydes. In this way the 1-azaspiro[4,5]decanes 7-9 were prepared in isolated yields of greater than 50% from their cyclohexanone precursors. It is significant that the 1-azaspirane formed from 4-t-butylcyclohexanone was predominantly a single isomer (92%)⁵, and the stereochemical assignment for 8 follows from the expectation that sigmatropic rearrangement would occur preferentially⁶ across the equatorial face of a cyclohexanone iminium ion intermediate. Azaspirane 7 was produced in lower yield (45%) from the direct² reaction of amino alcohol 3 and cyclohexanone, and was not formed to a significant extent when oxazolidine 6 was treated with 0.10 equiv of the acid catalyst. The imine prepared from cyclohexanone and 2-methoxy-2-methyl-3-butenamine also afforded azaspirane 7 (57% yield) when treated for 72 h in refluxing benzene with 1.0 equiv of *p*-toluenesulfonic acid monohydrate. Preliminary indications are that the use of ketones in the pyrrolidine synthesis of eq 1 may be limited to cyclohexanones, since 10 was formed in only low yield from 3-pentanone, and rearrangement of the 5-methyl-5-vinyloxazolidine derived from cyclopentanone afforded no recognizable 1-azaspiro [4,4]octane products.⁷

Treatment of oxazolidine 6 in refluxing benzene for 24 h with alkylating agents (1.0 equiv) afforded the 1-N-alkyl-1-azaspiro[4,5]decanes 11 in reasonable yield, together with smaller amounts of the nonalkylated azaspirane 7 (eq 2). Results are summarized in Table II.



The two-step procedure described here represents one of the most convenient constructions of the 1-azaspiro[4,5]decane ring system reported to date.⁸ Moreover, the ability to utilize 5-vinyloxazolidines in directed 2-azonia-[3,3]-sigmatropic rearrangement synthetic strategies² further expands the potential of this mild carbon-carbon bond forming method.

TABLE I. Preparation of 3-Acetylpyrrolidines According to Eq. 1.

OXAZOLINE		ACETYPYRROLIDINE	
	92% bp 75-5(3) ^a		69% ^b
	91% bp 140-5(3) ^a		70% ^b
	91% ^c bp 85(3) ^a		55% ^{b,d}
	86% ^c bp 85(5) ^a		65% ^b
	94% ^c bp 100-3(5) ^a		9% ^b

^a mm Hg. ^b Isolated bulb-to-bulb distillation: bath temperature: 80-120°C, 0.05 mm Hg. ^c A catalytic amount of p-toluenesulfonic acid was employed. ^d A mixture of four isomers.

TABLE II.

RX	Solvent	<u>11</u> ,%	<u>7</u> ,%
MeI	C ₆ H ₆	54	17
MeOTs	C ₆ H ₆	49	17
n-C ₈ H ₁₆ I	DMF	48	21
n-C ₆ H ₁₂ I	DMF	45	9

The following procedures are representative:

2-Methyl-2-vinyl-1-oxa-4-azaspiro[4,5]decane (6). A solution of cyclohexanone (2.95g, 30 mmol), 1-amino-2-methyl-3-buten-2-ol ⁴ (3.03g, 30 mmol), and 30 mL of benzene was heated at reflux for 2h with azeotropic removal of water. Concentration and distillation through a short Vigreux column afforded 5.0g (92%) of 6: bp 73-75°C(3 mm); IR (film) 3300, 1450, 1080, 920 cm⁻¹; ¹H-NMR (CDCl₃) δ5.1-5.8 (m, CH=CH₂), 2.97 (d, J=1.7Hz, CH₂NH), 1.6 (m, NH), 1.29 (s, CH₃).

3-Acetyl-1-azaspiro[4,5]decane (7). A solution of 6 (543 mg, 3.00 mmol), d-10-camphor-sulfonic acid (696 mg, 3.00 mmol), and 3 mL of anhydrous benzene was heated at reflux under N₂ for 3h. After the mixture cooled to room temperature, 3mL of 1 N NaOH was added and the amine product was isolated by ether extraction and dried (Na₂SO₄). Distillation (bulb-to-bulb; bath temperature 80-85°C; 0.05 mm) yielded 375 mg (69%) of pure 7: IR (film) 3370, 1710 cm⁻¹; ¹H-NMR(CDCl₃) δ2.17 (s, CH₃CO); ¹³C-NMR (CDCl₃) δ209.1 (C=O), 62.9 (C-5), 52.6 (C-2), 47.8 (C-3), 39.7, 38.2, 37.3, 29.2 (CH₃CO), 25.9, 23.9, 23.6.

Acknowledgment. This work was supported at the University of California by a grant from the National Institutes of Health (NS-12389).

References and Notes

- Camille and Henry Dreyfus Teacher-Scholar, 1976-81. Address correspondence to this author at the University of California.
- Overman, L.E.; Kakimoto, M. J. Am. Chem. Soc. 1979, 101, 1310.
- For a recent review of [3,3]-sigmatropic rearrangements of iminium salts see: Heimgartner, H.; Hansen, H.-J.; Schmid, H. in "Iminium Salts in Organic Chemistry," Part 2: Böhme, H.; Viehe, H.G., Ed.; Wiley: New York, 1979, pp 655-732.
- Prepared from isoprene oxide and NH₃: Mishra, A.; Rice, S.N.; Lwowski, W. J. Org. Chem. 1968, 33, 481. Murata, K.; Terada, A. Bull. Chem. Soc. Jpn. 1968, 41, 495.
- By ¹H-NMR analysis in the presence of Eu(FOD)₃.
- Cf. House, H.O.; Lubinkowski, J.; Good, J.J. J. Org. Chem. 1975, 40, 86.
- In both of these cases low molecular weight, water soluble, products were produced.
- For classical procedures see: Moffett, R.B. J. Am. Chem. Soc. 1957, 79, 3186. Hill, R.K. J. Org. Chem. 1957, 22, 830.

(Received in USA 15 June 1979)